

## The Role of Dihydrogen Oxide in Dental Practice

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### ABSTRACT

The role of dihydrogen oxide has recently been brought to the attention of the dental profession. This short article examines the origin, history, recent uses and future possibilities for this compound.

### INTRODUCTION AND LITERATURE REVIEW

Dihydrogen oxide is a common compound in the earth's atmosphere and in soils, rocks and other sedimentary deposits. It is formed as the reaction product of two common elements, hydrogen and oxygen and in nature it may be found with varying levels of trace elements such as cadmium, fluorine, molybdenum, manganese, chlorine and others. Its use in industrial processes is wide spread, such as in the cloth manufacturing industry and in food processing. It is extensively used in the production of electricity and in nuclear power plants. It can exist in three forms – gaseous, liquid and solid, depending on the temperature.

In the domestic situation, it is used as a cleaning agent, and also in the preparation of foodstuffs. A regular supply is a mark of how developed a country is on the UN international index of development.

More recently its role in medicine and dentistry has highlighted many of its useful properties. It can be used as a solvent, and as a cleaning agent in many clinical processes. Purified forms of this compound can be used in clinical medicine and various elements can be added to it such as sodium chloride to ensure isomolarity. In the administration of drugs it can be used to dissolve various compounds and ingested, it can be injected using a syringe and needle, or it can be used in a vapourised form such as in aerosols. In dental practice, its use in cleansing the oral cavity is most prominent, while the introduction of various dental cements relies on its addition to a powder to transform the components into a flowable material.

Antibacterial properties of dihydrogen oxide remain equivocal. In certain cases it can be useful in gross cleaning of the infected area to remove large visible amounts of contaminants, but as an antibiotic it is not particularly effective.

Dihydrogen oxide does have a number of side effects. If ingested in too large quantities it can result in excessive micturition and an increase in blood pressure. It forms a large proportion of alcoholic drinks and alcohol addiction is a problem in many countries. It has been found that alcoholics have a severe negative reaction to the prospect of ingesting significant quantities of this compound and rehabilitation is often necessary in these cases. Its presence in sea water is very high and shipwrecked sailors have died from drinking too much sea water.

All dentists should make use of this common material and ensure that their daily practice is geared towards the efficient use and disposal in the dental surgery.

### REFERENCES

1. Walker, J. (1999). The use of dihydrogen oxide in the production of ethanol derivatives. *The Journal of Distillation Research*. 47: 19-23.
2. Grant W. (2017). Awa' ye Sassenach bastards. Can't ye see I'm pished? *Research in Addiction*. 93: 12-1042.
3. Glenfiddich G. (1997). On the proportions of dihydrogen oxide in ethanol derivatives at low temperatures. *Archives of Fluid Dynamics*. 17: 4-9.